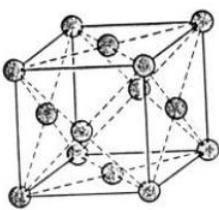
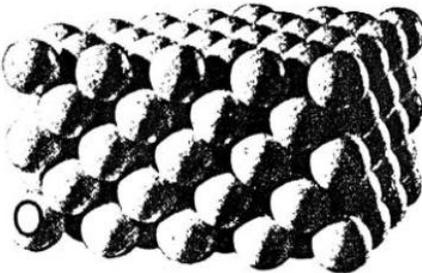


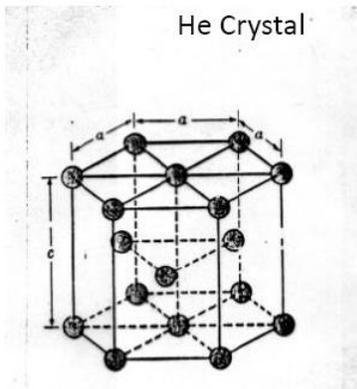
**Phys 155, Winter 2011, Homework 1, due Jan 13**

Each problem is worth 10 points.

- (a) Use the semi-classical Bohr-Sommerfeld quantization condition  $\oint p dq = nh$  (the integral is over an orbit corresponding to one period) to derive the energy level of a one dimensional simple harmonic oscillator,  $E_n = n\hbar\omega$ , where  $\omega$  is the natural frequency of the oscillator. Remember to treat everything with classical mechanics, except for the quantization condition itself. Use the solution  $x = A \cos(\omega t + \phi)$  to evaluate the integral. (b) One would generally say that the solution of (a),  $E_n = n\hbar\omega$ , although incorrect for small  $n$  values, is an excellent solution since we are dealing with semi-classical physics (thus high  $n$ ). Nevertheless, the Bohr-Sommerfeld quantization condition can be, and is often, modified to include a problem-dependent constant  $\gamma$ ,  $\oint p dq = (n + \gamma)h$ , to give the correct result even for small  $n$  values. Assuming  $n = 0, 1, 2, \dots$  what is the value of  $\gamma$  necessary for this problem?
- For each of the following three crystals, indicate a primitive [unit] cell, and indicate how many atoms there are per [unit] cell. Give brief explanations why.

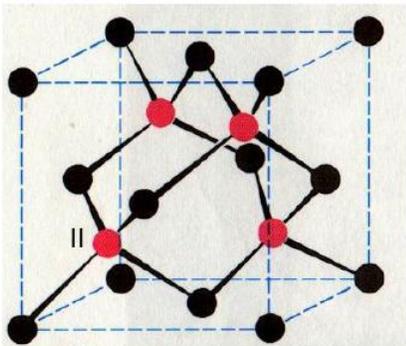
Ar Crystal



He Crystal



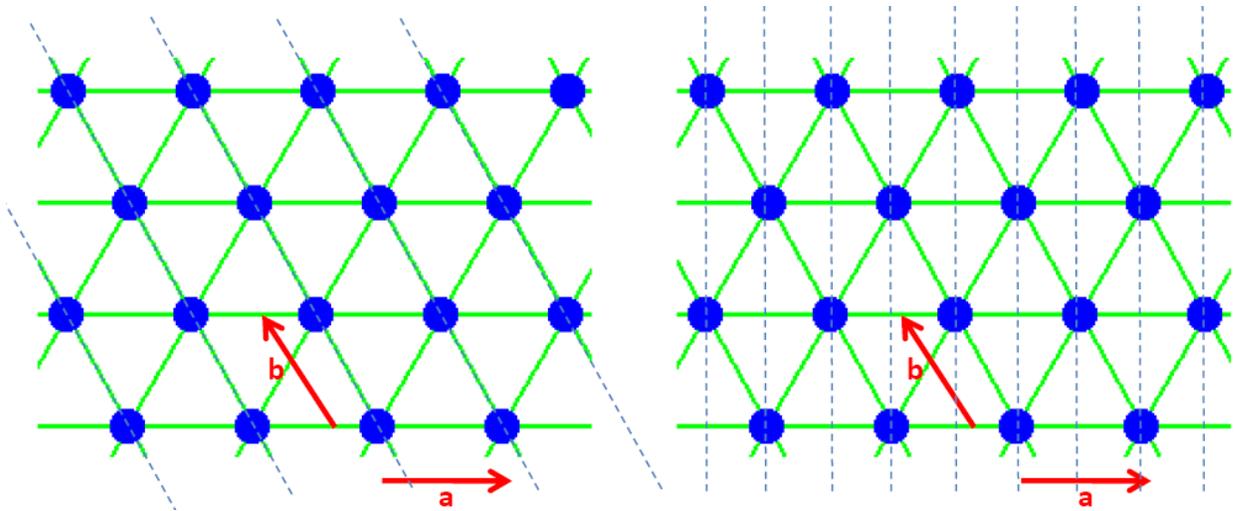


Diamond, if I=II=C (Carbon)  
 Similarly for Si, Ge  
 ZnS, if I=Zn, II=S  
 Similarly for GaAs, InAs, ...

- Consider the planes with indices (100), (110) and (001): the lattice is fcc, and the indices refer to the conventional cubic cell (with side  $a$ ). What are the indices of these planes

when referred to the primitive axes defined by the three primitive vectors:  $\vec{a} = \frac{1}{2}(\hat{x} + \hat{y})$ ,  $\vec{b} = \frac{1}{2}(\hat{y} + \hat{z})$ ,  $\vec{c} = \frac{1}{2}(\hat{z} + \hat{x})$ .

4. Find Miller indices for the following two sets of lattice lines (dashed lines) for the lattice given. Explain your answer.



5. Show that the  $c/a$  ratio of the ideal hexagonal closed-packed structure is  $\sqrt{\frac{8}{3}} = 1.633$ .